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## Electrochemical determination of glutathione

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### Abstract

Procedures were developed for determining glutathione by voltammetry and coulometric titration with electrogenerated oxidants using the biamperometric indication of the titration end-point. Possible mechanisms of the glutathione reaction with electrogenerated halogens are discussed. Microgram amounts of glutathione can be determined in model solutions with an RSD of 1-2%. The oxidation wave of glutathione in the voltammogram is observed at 0.95 V. At higher glutathione concentrations, the wave takes the shape of a peak. Glutathione concentration in the range between  $9.15 \times 10^{-5}$  and  $2.14 \times 10^{-3}$  M is a linear function of its oxidation wave height at a stationary platinum electrode in a 0.05 M H<sub>2</sub>SO<sub>4</sub> solution. The determination limit for glutathione is  $1.9 \times 10^{-5}$  M. The procedures for determining glutathione in human blood were proposed.

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